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Feeling the Pressure: Polymorphism, Particle Size and Morphology Control Through Nanoparticle Crystallization Using Controlled Expansion of Supercritical Solutions

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PURPOSE

so for nanoparticles.

METHODS

Poor bioavailability and low efficacy arising from

solubility and increase the dissolution rate of

The generation of crystalline nanoparticles can

increase the specific surface area of API particles

leading to higher dissolution rate whilst

maintaining the stability of the crystalline form

and avoiding the necessity for additional excipients, as is the case for an amorphous solid

dispersion. Controlling crystalline form,

morphology and particle size distribution (PSD) is

critical for any crystallization process, even more

The key methodology for this work is Controlled

Expansion of Supercritical Solution (CESS®)

technology [1], which is a particle production

method based on supercritical carbon dioxide

(scCO₂) without the addition of further solvents.

Atovaguone (ATO) and Fenofibrate (FEN) were

the chosen APIs for the study owing to the

presence of multiple polymorphic forms and their

predicted high solubility in scCO₂.

Fig. 1. The CESS® process

active pharmaceutical ingredients (APIs).

RESULTS CESS[®] process

poor solubility are amongst the most common reasons for drug failure. Converting crystalline material to the amorphous form is an alternative means of improving solubility but physical stability of amorphous forms is a drawback in its applicability. An advanced approach is to use crystalline manoparticles to improve drug

Fenofibrate: particle sizing

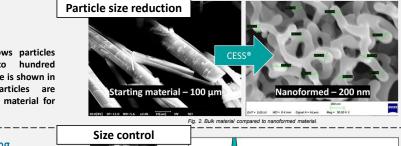
In the case of FEN, the PSD of Form I can be controlled merely by varying the temperature at which crystallization occurs. In this way, the PSD can be tailored to the requirement of the desired dosage form without the need for separate processes or solvent systems being developed.

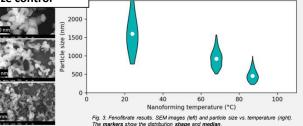
Atovaguone polymorphism

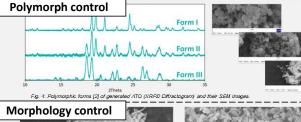
Through careful control of the CESS process, it is possible to manipulate the polymorphic form of ATO generated from the crystallization process using scCO₂, between Forms I, II and III, without the need for additional solvents.

Atovaquone morphology

Figure 5 demonstrates that the morphology of the nanoparticles of thermodynamically stable Form III can be modified from the lath preference of the larger particles to the more processable cubic shape.







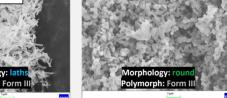


Fig. 5. Morphology control, generating either rods or spherical particles of the same polymorphic form.

CONCLUSIONS

By controlling the temperature of crystallization, it is possible to vary the PSD of crystalline FEN Form I from 0.5 to 1.5 μ m in order to tailor the particles for a particular dosage form.

Through manipulation of the CESS process, it is possible to not only generate nanoparticles of metastable ATO Forms I and II but also to produce nanoparticles of the thermodynamically stable Form III with a modified, and more processable, morphology.

The versatility of CESS is truly remarkable, using the same process and the same solvent system it is possible to control the particle size, polymorphism and morphology merely by varying the process parameters.

Property	Benefit for Drug Development
Nanoparticles	Higher dissolution rate Higher bioavailability
Size control	Tuning for formulation
Polymorph control	Higher solubility Higher stability Intellectual property
Morphology control	Easier formulation

▲ Table 1. Effect of the physical properties of the powde

References

 Pessi, J., Lassila, I., Meriläinen, A., Räikkönen, H., Hæggström, E., & Yliruusi, J. (2016). Controlled expansion of supercritical solution: a robust method to produce pure drug nanoparticles with narrow size-distribution. *Journal of pharmaceutical sciences*, 105(8), 2293-2297.

[2] Tarur, Venkataswubramanian. "Novel polymorphs of atovaquone and process of." U.S. Patent Application No. 10/569,036.

